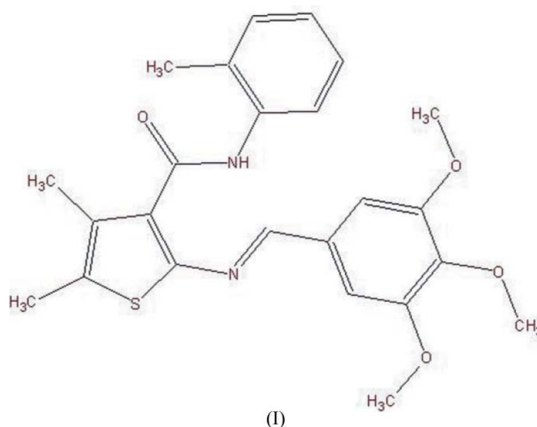


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Key indicators

Single-crystal X-ray study
T = 290 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.045
wR factor = 0.123
Data-to-parameter ratio = 16.1For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.4,5-Dimethyl-N-(2-methylphenyl)-2-[[*(1E)*-(3,4,5-trimethoxyphenyl)methylene]amino]-thiophene-3-carboxamideThe conformation of the title molecule, $\text{C}_{24}\text{H}_{26}\text{N}_2\text{O}_4\text{S}$, is influenced by intramolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonding, and $\pi-\pi$ interactions link the molecules into centrosymmetric dimers. The crystal packing is further stabilized by weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.Received 28 December 2006
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Comment

The title compound, (I), shows promising antibacterial activity against *subtilis aureus* and *Escherichia coli*, comparable with the activity of ampicillin (Mohan & Saravanan *et al.*, 2002, 2003). We present here the crystal structure of the title compound, (I).In (I) (Fig. 1), all bond lengths and angles show normal values (Allen *et al.*, 1987). An intramolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bond (Table 1) forms a pseudo-six-membered ring with graph-set notation $S(6)$ (Bernstein *et al.*, 1995) and influences the molecular conformation. The *o*-toluidine is *gauche* to the thiophene ring, the dihedral angle being $59.24(2)^\circ$. The geometric parameters of the methoxy groups correspond to those observed in methoxybenzene derivatives (Fun *et al.*, 1997).The relatively short $\text{Cg1}\cdots\text{Cg2}^{\text{iii}}$ distance of $3.683(3) \text{ \AA}$ (Cg1 and Cg2 are centroids of the $\text{S1/C1}-\text{C4}$ and $\text{C16}-\text{C21}$ rings, respectively) shows the presence of $\pi-\pi$ interactions which link the molecules into centrosymmetric dimers [symmetry code: (iii) $-x + 2, -y + 1, -z$]. The crystal packing (Fig. 2) is further stabilized by weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 1).

Experimental

The title compound was synthesized using the Gewald reaction (Gewald *et al.*, 1966). *o*-Cyanotoluidine (0.04 mol) was refluxed with

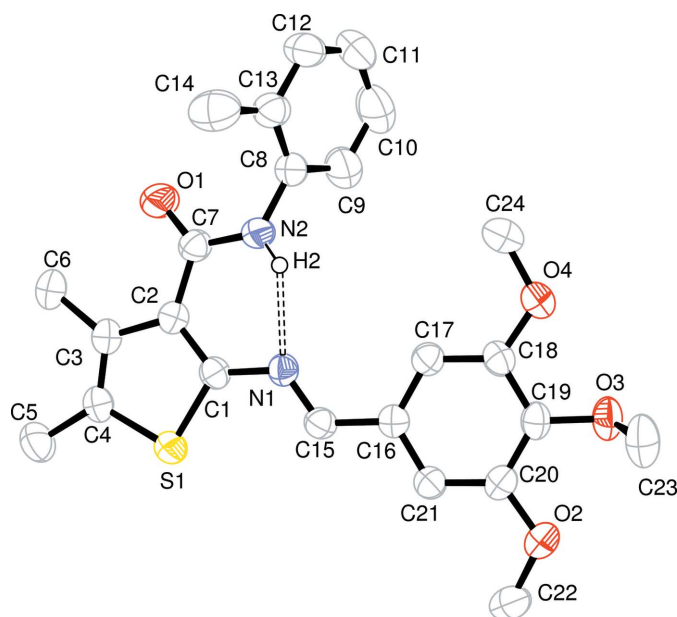


Figure 1
The molecular structure of (I), showing the atomic labeling and 50% probability displacement ellipsoids. The double dashed line denotes the N—H...N hydrogen bond. All H atoms, except H2, have been omitted for clarity.

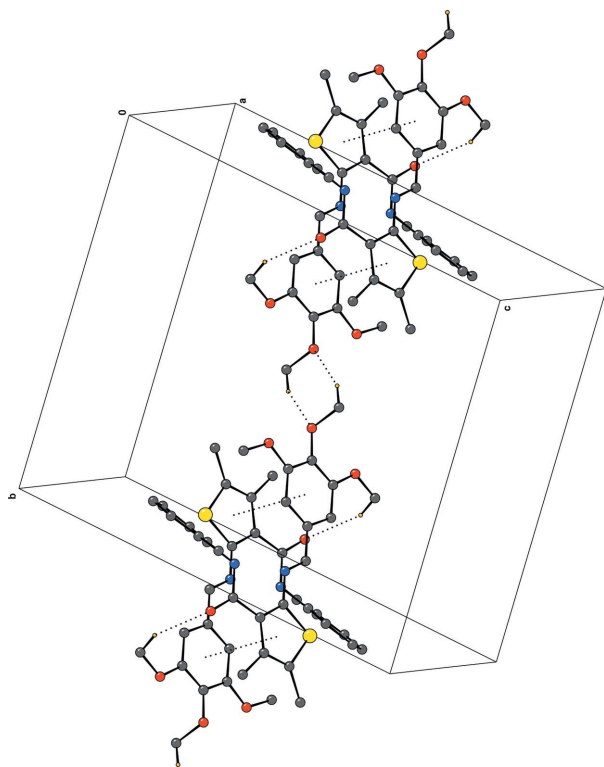


Figure 2
A portion of the crystal packing of (I). The dotted lines denote π – π interactions and C—H...O hydrogen bonds. H atoms not involved in hydrogen bonds have been omitted for clarity.

ethyl methyl ketone in the presence of sulfur (0.04 mol), dimethylamine (4 ml) and ethanol (40 ml) at 313–323 K for 1 h. The product was then reacted with 3,4,5-trimethoxybenzaldehyde in an equimolar ratio in the presence of ethanol, which yielded the title compound.

This was then purified and crystallized from dimethylformamide and ethanol (1:2) by slow evaporation, yielding orange block-shaped crystals.

Crystal data

$C_{24}H_{26}N_2O_4S$
 $M_r = 438.54$
Monoclinic, $P2_1/n$
 $a = 7.1802$ (15) Å
 $b = 16.064$ (3) Å
 $c = 19.354$ (4) Å
 $\beta = 90.123$ (3)°
 $V = 2232.3$ (8) Å³

$Z = 4$
 $D_x = 1.305$ Mg m⁻³
Mo $K\alpha$ radiation
 $\mu = 0.18$ mm⁻¹
 $T = 290$ (2) K
Block, orange
0.30 × 0.25 × 0.25 mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1997)
 $T_{\min} = 0.909$, $T_{\max} = 0.957$

17378 measured reflections
4661 independent reflections
3427 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.043$
 $\theta_{\text{max}} = 27.3^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.123$
 $S = 1.00$
4661 reflections
290 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0751P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.21$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N2-H2\cdots N1$	0.86	2.10	2.737 (2)	130
$C22-H22C\cdots O1^i$	0.96	2.57	3.459 (3)	155
$C23-H23B\cdots O3^{ii}$	0.96	2.56	3.424 (3)	150

Symmetry codes: (i) $-x + 1, -y, -z + 1$; (ii) $-x + 1, -y - 1, -z + 1$.

All H atoms were located in a difference Fourier map, placed in idealized positions ($C-H = 0.93$ – 0.96 and $N-H = 0.86$ Å) and refined as riding, with $U_{\text{iso}}(H) = 1.2$ or 1.5 times U_{eq} (parent atom).

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SIR92 (Altomare *et al.*, 1993); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and CAMERON (Watkin *et al.*, 1993); software used to prepare material for publication: PLATON (Spek, 2003).

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References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Altomare, A., Casciarano, G., Giacovazzo, C. & Guagliardi, A. (1993). *J. Appl. Cryst.* **26**, 343–350.

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Bruker (2000). *SMART* and *SAINTE*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Fun, H.-K., Chinnakali, K., Sivakumar, K., Sam, T.-W. & How, S.-E. (1997). *Acta Cryst. C* **53**, 1859–1862.
- Gewald, K., Schinke, E. & Botcher, H. (1966). *Chem. Ber.* **99**, 94–100.
- Mohan, S. & Saravanan, J. (2002). *Indian J. Heterocycl. Chem.* **12**, 87–88.
- Mohan, S. & Saravanan, J. (2003). *Asian J. Chem.* **15**, 67–70.
- Sheldrick, G. M. (1997). *SADABS* and *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Watkin, D. M., Pearce, L. & Prout, C. K. (1993). *CAMERON*. Chemical Crystallography Laboratory, University of Oxford, England.